

## AN EPIDITHIAPIPERAZINEDIONE ANTIVIRAL AGENT

FROM ASPERGILLUS TERREUS

by P. A. Miller, P. W. Trown, W. Fulmor,

G. O. Morton, and J. Karliner<sup>1</sup>

Lederle Laboratories Division

American Cyanamid Company

Pearl River, New York 10965

Received August 30, 1968

Culture filtrates of Aspergillus terreus were found to be active in an antiviral screen designed to detect specific inhibitors of viral RNA synthesis (Trown et al., 1967). One of the active components appeared to be gliotoxin whose mode of action we recently discussed (Miller et al., 1968). We now describe the isolation and preliminary characterization of another active metabolite, designated LL-S88 $\alpha$ , which also belongs to the epidithia-piperazinedione class of natural products. The determination of the structure of LL-S88 $\alpha$  by x-ray crystallography is described elsewhere (Cosulich et al., 1968)<sup>2</sup>.

LL-S88 $\alpha$  is produced by many strains of Aspergillus terreus when this organism is grown on a synthetic medium. Isolation of LL-S88 $\alpha$  from culture filtrates was monitored using an assay based on inhibition of incorporation of uridine-5-<sup>3</sup>H into viral RNA, an in vivo test involving a Coxsackie A21 virus infection of mice and by silica gel thin-layer chromatography. LL-S88 $\alpha$  was visualized on thin-layer chromatograms by its uv absorption, charring with sulfuric acid or spraying with a sodium azide-iodine reagent (Feigl, 1956);

---

<sup>1</sup> Now with Geigy Research Labs, Ardsley, N. Y. 10502.

<sup>2</sup> After completion of the x-ray studies, Dr. N. Neuss of the Eli Lilly Co. kindly supplied us with samples of acetylaranotin (I) and bisdethio-di-(methylthio)-acetylaranotin (II) (Nagarajan et al., 1968). We find LL-S88 $\alpha$  identical to I and LL-S88 $\beta$  identical to II.

the latter has been reported to be very sensitive in detecting the epidithia-piperazinedione class of compounds (Taylor, 1966).

Culture filtrate of Aspergillus terreus NRRL-3319 was extracted with chloroform and the chloroform extract concentrated to a dark thick syrup which was dried in vacuo. A fraction which did not dissolve in a small volume of dry chloroform was discarded and after evaporation of the solvent, the residual solids were dissolved in hot ethanol. A crystalline product which precipitated from the ethanol was further purified on a silica gel column developed with chloroform. The resulting product was recrystallized from ethanol to give pale yellow tetragonal bipyrimidal crystals, m.p. 215 - 230° dec. Anal. Calcd. for  $C_{22}H_{20}N_2O_8S_2$ : C, 52.38; H, 4.00; N, 5.56; O, 25.39; S, 12.69. Found: C, 53.13; H, 4.07; N, 5.41; O, 24.11; S, 12.37. The high resolution mass spectrum contained an ion with  $m/e = 440.1221$  which appears to have the composition  $C_{22}H_{20}N_2O_8$  (calc. 440.1218) and to be derived from the molecular ion by loss of  $S_2$  (Bose et al., 1968). Uv ( $CH_3OH$ ) end absorption with shoulders at 270 ( $\epsilon 1800$ ) and 225  $m\mu$  ( $\epsilon 10,200$ ); ir (KBr) 5.73, 5.83, 7.32, 7.42, 8.14, 8.76, 9.62, and 9.79  $\mu$ ; nmr ( $CDCl_3$ ) 2.00, 3.00, 4.00, 4.55, 5.07, 5.60 and 6.25  $\delta$ .

A second metabolite, designated LL-S88 $\beta$ , was isolated from the ethanol mother liquors of the LL-S88 $\alpha$  crystallization. LL-S88 $\beta$  did not catalyze the reaction between sodium azide and iodine (Feigl, 1956) nor did it inhibit the synthesis of viral RNA. Its isolation was therefore followed by thin-layer chromatography where the compound was detected by its uv absorption and staining with  $I_2$ . Silica gel chromatography followed by repeated recrystallization from ethanol yielded pure LL-S88 $\beta$  as white platelets, m.p. 215 - 236° dec. Anal. calcd. for  $C_{24}H_{26}N_2O_8S_2$ : C, 53.91, H, 4.91; N, 5.24; S, 11.97. Found C, 53.63; H, 4.95; N, 5.21; S, 12.10. The high resolution mass spectrum contained an ion with  $m/e = 487.1109$  which has the composition  $C_{23}H_{23}N_2O_8S$  and is presumed to arise from the molecular ion by loss of  $-SCH_3$ . A second loss of  $-SCH_3$  would result in an ion with  $m/e$  440 as was observed. The ms of LL-S88 $\alpha$  and LL-S88 $\beta$  were similar in the regions below  $m/e$  440. Uv ( $CH_3OH$ ) end absorption

with shoulders at 255 ( $\epsilon$  1900) and 220 m $\mu$  ( $\epsilon$  14,100); ir 5.74, 5.96, 7.30, 8.21, 8.75, 8.86, 9.72  $\mu$ ; nmr (CCl<sub>3</sub>), 2.06, 2.25, 3.02, 4.68, 5.18, 5.79, 6.29, 6.56  $\delta$ .

Analysis of nmr, ms, ir and uv data did not unambiguously suggest a structure for either component. The data did however suggest that LL-S88 $\alpha$  is a symmetrically substituted epidithiapiperazinedione and that LL-S88 $\beta$  is its dithiomethyl ether derivative. The interrelationship is supported by the lack of reactivity of LL-S88 $\beta$  with the sodium azide-iodine reagent (Feigl, 1956), and was confirmed by the conversion of LL-S88 $\alpha$  to LL-S88 $\beta$  by reduction with sodium borohydride and methylation with methyl iodide in chloroform-methanol. A similar conversion in the sporidesmin series of compounds has been reported (Rahman *et al.*, 1967).

LL-S88 $\alpha$  is active in tissue culture against strains of rhino-, Coxsackie, polio- and parainfluenza viruses, and protects mice against lethal infections produced with Coxsackie A21 or influenza B/Md viruses<sup>3</sup>. No antiviral activity has been observed for LL-S88 $\beta$ . LL-S88 $\alpha$  completely blocks viral RNA synthesis at levels which are without effect on cellular RNA synthesis<sup>3</sup>. We believe that this specific action is probably the basis for its antiviral activity.

#### References

- Bose, A. K., Das, K. G., Funke, P. T., Kugajevsky, I., Shukula, O. P., Khanchandani, K. S., and Suhadolnik, R. J., *J. Amer. Chem. Soc.*, **90**, 1038 (1968).  
Cosulich, D. B., Nelson, N. R., and van den Hende, J. H., *J. Amer. Chem. Soc.*, in press.  
Feigl, F., "Spot Tests in Organic Analysis," 5th Ed., Elsevier Pub. Co., Amsterdam, 1956, p. 230.  
Miller, P. A., Milstrey, K. P., and Trown, P. W., *Science*, **159**, 431 (1968).  
Nagarajan, R., Huckstep, L. L., Lively, D. H., DeLong, D. C., Marsh, M. M., and Neuss, N., *J. Amer. Chem. Soc.*, **90**, 2980 (1968).  
Rahman, R., and Taylor, A., *Chem. Comm.*, 1032 (1967).  
Taylor, A., "Biochemistry of Some Foodborne Microbial Toxins," R. I. Mateles and G. N. Wogan, Eds., The M. I. T. Press, Cambridge, Mass., 1966, p. 74.  
Trown, P. W., Brindley, K. P., and Miller, P. A., *Proc. Int. Cong. Chemotherapy Vth*, Vol. II (1), 7 (1967).

<sup>3</sup> The biological properties will be described elsewhere.